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## **CHAPTER ASSESSMENT**

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# Reviewing Vocabulary

Match the definition in Column A with the term in Column B.

#### Column A

- 1. A reaction in which a compound breaks down into two or more elements or new compounds
- 2. A number written in front of a chemical formula
- 3. A solid produced during a chemical reaction in a solution
- 4. A solution in which the solvent is water
- 5. A statement that uses chemical formulas to show the identities and relative amounts of the substances involved in a chemical reaction
- what type of joins are present in s 6. An equation that shows all of the particles in solution as they actually exist
  - 7. Substance dissolved in a solution.
- 8. An equation that includes only the particles that participate in the reaction
  - 9. An ion that is present but does not participate in a reaction
    - 10. A reaction in which oxygen combines with a substance and releases heat and light energy
- 11. A reaction in which the atoms of one element replace the atoms of another element in a compound a, complete tonic b. notaouic
  - 12. A reaction involving the exchange of positive ions between two compounds dissolved in water
    - \_ 13. The process by which the atoms of one or more substances are rearranged to form different substances
- 14. A starting substance in a chemical reaction
  - 15. A substance formed during a chemical reaction
  - **16.** A reaction in which two or more substances react to produce a single product
- \_ 17. The most plentiful substance in a solution.

### Column B

- a. aqueous solution
- **b.** chemical equation
- c. chemical reaction
- d. coefficient
- e. combustion reaction has taning the salt to thee roll
  - f. complete ionic equation
  - g. decomposition reaction
  - h. double-replacement
  - i. net ionic equation
  - precipitate
  - k. product
  - I. reactant
  - m. single-replacement reaction
  - **n.** spectator ion
  - o. synthesis reaction
  - p. solute
  - q. solvent