

Chemistry

Due
May 18

41. calcium oxide + diphosphorus pentoxide \rightarrow calcium phosphate.
42. chromium + oxygen \rightarrow
43. sodium + water \rightarrow
44. calcium carbonate + hydrochloric acid \rightarrow
45. calcium hydroxide + phosphoric acid (H_3PO_4) \rightarrow
46. sodium carbonate + nitric acid \rightarrow
47. aluminum hydroxide + sulfuric acid \rightarrow
48. sodium sulfite + sulfuric acid \rightarrow
49. copper + sulfuric acid \rightarrow copper(II) sulfate + water + sulfur dioxide(g).
50. calcium hydroxide + ammonium sulfate \rightarrow calcium sulfate + water + ammonia(g).

Balance The above equations

The first reactant is 10g and the second reactant is 7g

Calculate all products Remember limited reaction

Explain all work

Show all work